Acta Crystallographica Section E

## Structure Reports

Online
ISSN 1600-5368

## Xiao-Kang Ai, Cai-Feng Bi,* Yu-Hua Fan, Xia Zhang and Xue-Tao He

College of Chemistry and Chemical Engineering, Ocean University of China, Qingdao 266003, People's Republic of China

Correspondence e-mail: bcfeng@ouc.edu.cn

## Key indicators

Single-crystal X-ray study
$T=293 \mathrm{~K}$
Mean $\sigma(\mathrm{C}-\mathrm{C})=0.003 \AA$
$R$ factor $=0.044$
$w R$ factor $=0.134$
Data-to-parameter ratio $=16.7$
For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

[^0]
## 2-[(2-Aminophenylimino)(phenyl)methyl]-4-chlorophenol

The title compound, $\mathrm{C}_{19} \mathrm{H}_{15} \mathrm{ClN}_{2} \mathrm{O}$, a tridentate Schiff base, has been synthesized and structurally characterized by IR spectroscopy and X-ray structure analysis. The hydroxy and imino groups are involved in a resonance-assisted $\mathrm{O}-\mathrm{H} \cdots \mathrm{N}$ hydrogen bond with $\mathrm{O} \cdots \mathrm{N}=2.517$ (2) Å. Intermolecular $\mathrm{O}-$ $\mathrm{H} \cdots \mathrm{N}$ and $\mathrm{N}-\mathrm{H} \cdots \mathrm{O}$ hydrogen bonds create centrosymmetric dimers in the crystal packing.

## Comment

2-[(2-Aminophenylimino)(phenyl)methyl]-4-chlorophenol has been widely investigated as a useful unsymmetrical tetradentate Schiff base ligand for the preparation of metal complexes (Bi \& Fan, 2004; Zhu et al., 2004; Kong \& Ding, 1999; Atkins et al., 1985). In the course of the synthesis of such a complex, single crystals of the compound, (I), were obtained. The crystal structure of (I) is reported here (Fig. 1).

(I)

In the molecular structure of (I), the hydroxy group is involved in an intramolecular $\mathrm{O}-\mathrm{H} \cdots \mathrm{N}$ hydrogen bond (Fig. 1, Table 1), through which atoms O1, H1, N2, C7, C14 and C 15 form a six-membered ring. The $\mathrm{C} 7=\mathrm{N} 2[1.298$ (2) $\AA]$, $\mathrm{C} 7-\mathrm{C} 14 \quad[1.480(2) \AA]$ and $\mathrm{O} 1-\mathrm{C} 15$ [1.348 (2) $\AA$ ] bond lengths are shorter than normal $\mathrm{C}=\mathrm{N}(1.32 \AA), \mathrm{C}-\mathrm{C}(1.54 \AA)$ and $\mathrm{O}-\mathrm{C}(1.44 \AA)$ bonds; Allen et al., 1987). The C14=C15 bond length [1.419 (2) $\AA$ ] is longer than a normal $\mathrm{C}=\mathrm{C}$ bond ( $1.40 \AA$ in benzene). These differences in bond lengths are related to a resonance-assisted hydrogen bond (RAHB) acting between molecules with conjugated multiple $\pi$ bonds (Gilli et al., 2000).

In the crystal packing of (I), there are centrosymmetric dimers generated by $\mathrm{O}-\mathrm{H} \cdots \mathrm{N}$ and $\mathrm{N}-\mathrm{H} \cdots \mathrm{O}$ hydrogen bonds (Fig. 2 and Table 1).


Figure 1
The structure of the title compound, with $30 \%$ probability displacement ellipsoids and the atom-numbering scheme. The intramolecular hydrogen bond is shown as a dashed line.

## Experimental

The title compound, (I), was prepared according to the literature procedure of $\mathrm{Bi} \&$ Fan (2004). A mixture of 2-[[(2-amino-phenyl)imino]phenylmethyl]-4-chlorophenol ( 1.29 g ) dissolved in heated $n$-butanol $(25 \mathrm{ml})$ and cadmium acetate dihydrate $(1.07 \mathrm{~g})$ dissolved in methanol ( 25 ml ) was heated under reflux for 4 h . After cooling, the resulting solution was filtered and kept in air. After slow evaporation of the solvent over several weeks, large orange-red crystals of (I) formed at the bottom of the vessel. Single crystals suitable for X-ray analysis were isolated, washed twice with methanol and dried in air.

## Crystal data

$\mathrm{C}_{19} \mathrm{H}_{15} \mathrm{ClN}_{2} \mathrm{O}$
$M_{r}=322.78$
Triclinic, $P \overline{1}$
$a=8.5630$ (17) A
$b=9.4940(19) \AA$
$c=11.069(2) \AA$
$\alpha=68.96(3)^{\circ}$
$\beta=84.84(3)^{\circ}$
$\gamma=77.75(3)^{\circ}$

## Data collection

Enraf-Nonius CAD-4
diffractometer
diffractometer
$\omega$ scans
Absorption correction: none
3790 measured reflections
3549 independent reflections

## Refinement

Refinement on $F^{2}$
$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.044$
$w R\left(F^{2}\right)=0.134$
$S=1.02$
3549 reflections
213 parameters
H-atom parameters constrained
$V=820.7(3) \AA^{3}$
$Z=2$
$D_{x}=1.306 \mathrm{Mg} \mathrm{m}^{-3}$
Mo $K \alpha$ radiation
$\mu=0.24 \mathrm{~mm}^{-1}$
$T=293$ (2) K
Block, orange-red
$0.25 \times 0.20 \times 0.18 \mathrm{~mm}$

2455 reflections with $I>2 \sigma(I)$

$$
R_{\mathrm{int}}=0.012
$$

$\theta_{\text {max }}=27.0^{\circ}$
3 standard reflections every 100 reflections intensity decay: none

$$
\begin{aligned}
& \begin{array}{l}
w=1 /\left[\sigma^{2}\left(F_{\mathrm{o}}^{2}\right)+(0.0738 P)^{2}\right. \\
\quad+0.1224 P] \\
\quad \text { where } P=\left(F_{\mathrm{o}}^{2}+2 F_{\mathrm{c}}^{2}\right) / 3 \\
(\Delta / \sigma)_{\max }<0.001 \\
\Delta \rho_{\max }=0.26 \mathrm{e} \AA^{-3} \\
\Delta \rho_{\min }=-0.28 \text { e } \AA^{-3} \\
\text { Extinction correction: } S H E L X L 97 \\
\quad \text { (Sheldrick, 1997) } \\
\text { Extinction coefficient: } 0.025
\end{array} \text { (5) }
\end{aligned}
$$



Figure 2
The crystal packing of (I), viewed along the $c$ axis. Intermolecular hydrogen bonds are shown as dashed lines.

Table 1
Hydrogen-bond geometry ( $\AA^{\circ},{ }^{\circ}$ ).

| $D-\mathrm{H} \cdots A$ | $D-\mathrm{H}$ | $\mathrm{H} \cdots A$ | $D \cdots A$ | $D-\mathrm{H} \cdots A$ |
| :--- | :--- | :--- | :--- | :--- |
| $\mathrm{~N} 1-\mathrm{H} 1 A \cdots \mathrm{O} 1^{\mathrm{i}}$ | 0.86 | 2.31 | $3.035(2)$ | 142 |
| $\mathrm{O} 1-\mathrm{H} 1 \cdots \mathrm{~N} 2$ | $0.95(3)$ | $1.64(3)$ | $2.517(2)$ | $153(3)$ |

Symmetry code: (i) $-x+2,-y,-z+1$.
All H atoms were placed in geometrically idealized positions and constrained to ride on their parent atoms, with $\mathrm{N}-\mathrm{H}=0.86 \AA, \mathrm{C}-$ $\mathrm{H}=0.93 \AA$ and $\mathrm{O}-\mathrm{H}=0.95(3) \AA$, and with $U_{\text {iso }}(\mathrm{H})=1.2 U_{\text {eq }}(\mathrm{C})$ for the aromatic H atoms or $1.2 U_{\text {eq }}$ (parent) for the other atoms.

Data collection and cell refinement: CAD-4 Software (EnrafNonius, 1989); data reduction: NRCVAX (Gabe et al., 1989); structure solution: SHELXS97 (Sheldrick, 1997); structure refinement: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL/PC (Sheldrick, 1990); software used to prepare material for publication: WinGX (Farrugia, 1999).

## References

Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. \& Taylor, R. (1987). J. Chem. Soc. Perkin Trans. 2, pp. S1-19.

Atkins, R., Brewer, G., Kokot, E., Mockler, G. M. \& Sinn, E. (1985). Inorg. Chem. 24, 127-34.
Bi, C.-F. \& Fan, Y.-H. (2004). Xiyou Jinshu, 28, 699-702. (In Chinese.)
Enraf-Nonius. (1989). CAD-4 Software. Version 5.0. Enraf-Nonius, Delft, The Netherlands.
Farrugia, L. J. (1999). J. Appl. Cryst. 32, 837-838.
Gabe, E. J., Le Page, Y., Charland, J.-P., Lee, F. L. \& White, P. S. (1989). J. Appl. Cryst. 22, 384-387.
Gilli, P., Bertolasi, V., Ferretti, V. \& Gilli, G. (2000). J. Am. Chem. Soc. 122, 10405-10417.
Kong, F.-R. \& Ding, C.-Y. (1999). Huaxue Shijie, 40, 636-638. (In Chinese.)
Sheldrick, G. M. (1990). SHELXTL/PC. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.
Sheldrick, G. M. (1997). SHELXL97 and SHELXS97. University of Göttingen, Germany.
Zhu, B.-X., Ruan, W.-J., Yuan, R.-J., Cao, X.-H. \& Zhu, Z.-A. (2004). Yingyong Ниахие, 21, 1046-1050. (In Chinese.)


[^0]:    © 2006 International Union of Crystallography All rights reserved

